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## Research Article

# Optimization of Sintering Conditions for Improvement of Mechanical Property of $\alpha$ -Tricalcium Phosphate Blocks

### Abstract

Bioactive ceramic materials have been under research as bone substitute for several decades. To repair the high-load bearing bones, mainly cortical bones, there is a need for the substitute to possess comparable mechanical strength to cortical bone, of which the compressive strength ranges between 100 and 230 MPa. Two prevailing bone repairing material,  $\beta$ -tricalcium phosphate ( $\beta$ -TCP,  $\beta$ -Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>) and hydroxyapatite (HAp: Ca<sub>10</sub>(PO<sub>4</sub>)<sub>6</sub>(OH)<sub>2</sub>) have been widely researched and sintered into dense blocks to meet the mechanical requirements.  $\alpha$ -tricalcium phosphate ( $\alpha$ -TCP,  $\alpha$ -Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>), a high temperature polymorph of  $\beta$ -TCP, received relatively less attention and  $\alpha$ -TCP dense sintered blocks have not been reported yet. In this research, we fabricated  $\alpha$ -TCP dense blocks by sintering under various temperatures (1150-1400 °C) and the highest compressive strength was around 230 MPa. Intermediate porous blocks (porosity: 33%) were also fabricated from mixed powder of  $\alpha$ -TCP and starch. *In vitro* properties variation of the intermediate porous blocks were investigated by soaking samples in simulated body fluid (SBF) and the compressive strength was maintained above 100 MPa after soaking for 14 d.

## Introduction

Calcium orthophosphates materials have been under research and applied to clinical therapy for several decades. In terms of different types of human bones, various bone substitutes have been developed for the corresponding bones [1-3]. As is well known, human bones consist of two different types of bones, cortical bone and trabecular bone. The hard outer layer of bone is cortical bone (also referred as compact bone), which mainly exists in the shafts of long bones and as a shell around trabecular bone, containing around 80% of bone mass and facilitating bone's main functions. Trabecular bone (also referred as cancellous or spongy bone) can be found in the ends of long bones, in vertebrae and in flat bones like the pelvis. These two types of bones are classified on the basis of porosity and the unit microstructure. According to a variety of research results, cortical bone has a porosity ranging between 2% and 27% [4,5], while trabecular bone is much more porous and the porosity can be as large as 95% [6]. As for the mechanical properties of the bones, trabecular bone has been reported to exhibit a compressive strength of 2-10 MPa [7], and cortical bone possesses much higher compressive strength, ranging from 100-230 MPa [8].

Among a variety of calcium phosphates, there are three materials attracting great attention as bone repairing materials, referred as  $\alpha$ -tricalcium phosphate ( $\alpha$ -TCP,  $\alpha$ -Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>),  $\beta$ -tricalcium phosphate ( $\beta$ -TCP,  $\beta$ -Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>) and hydroxyapatite (HAp: Ca<sub>10</sub>(PO<sub>4</sub>)<sub>6</sub>(OH)<sub>2</sub>). Macroporous scaffolds that separately consist of these three types of materials attempting to repair porous bone have been reported in plenty of papers [9-11] and both of porosity and mechanical strength were comparable to those of trabecular

bone. However, when it comes to the cortical bone, requirements of repairing material become more complicate. Cortical bone mainly exists in the high load-bearing bones, like tibia and fibula. To repair the damaged sites in these bones, a compressive strength above 150 MPa of the repairing scaffold is generally demanded [12]. Meanwhile, as a candidate to repair bone, the substitute should be porous, or able to provide porosity after implantation [13]. Nevertheless, ceramic scaffolds get quite brittle when they become porous, making it difficult to reach a balance between the mechanical property and the porosity. More crucially, rapid return of function (formation of new bones) is also a requisite evaluation criteria for bone substitutes, which could bring the patients back to their normal daily life as quickly as possible, minimizing the negative influence to their quality of life. Based on the above three factors, mechanical strength, moderate porosity, biodegradability and osteoconduction are pivotal properties required for an optimal cortical bone substitute.

Among the three materials,  $\beta$ -TCP and HAp have already been widely used, both in macroporous and dense forms [14,15]. Some previous research has reported dense sintered blocks separately fabricated from  $\beta$ -TCP and HAp, of which the porosity was less than 1% and the compressive strength was far higher than that of human cortical bone [8]. Recently, H.O. Mayr et al. [16], reported an intermediate microporous  $\beta$ -TCP scaffold, which has less porosity (around 43.5%) than the macro porous scaffold and the pore size is merely around 5.40  $\mu$ m. After seeded with chondrocytes and implanted into a sheep model for one year, 80% of the scaffold was replaced by new bone. This research provides us a new possible approach to reach a balance between the mechanical strength and porosity suggesting that it may be feasible to fabricate relatively dense

scaffolds that match both the mechanical properties and porosity of human cortical bone, and in spite the pores are micro porous, the ability to be resorbed and replaced by human body may not be hindered.

As  $\alpha$ -TCP is relatively newly discovered, it is receiving growing attention because of  $\alpha$ -TCP has higher dissolution rate than  $\beta$ -TCP. Besides  $\alpha$ -TCP transforms easily into HAp by the reaction with  $H_2O$ . So,  $\alpha$ -TCP can be applied to bone cements or an additive in bone cements, although not investigated and applied as widely as the other two materials. The priority of  $\alpha$ -TCP to  $\beta$ -TCP or HAp has been reported by several research groups. In regard to the *in vitro* properties, at 37 °C and physical pH (7.2 ~ 7.4),  $\alpha$ -TCP releases relatively more Ca and P and the concentration ranks in the order of  $\alpha$ -TCP >  $\beta$ -TCP >> HAp [10,17], implying better ability of providing requisite resources for bone formation. Hisham Rojebani et al. [18], conducted an *in vivo* experiment of these three materials and confirmed that although  $\alpha$ -TCP,  $\beta$ -TCP and HAp were osteoconductive and successfully acted as space maintainer for bone formation when applied to a bone defect and  $\alpha$ -TCP showed the advantage of higher rate of degradation and more bone formation was observed. Both *in vitro* and *in vivo* tests manifested the better potential of  $\alpha$ -TCP to be resorbed by human body, whereby the return-of-function process may be promoted, especially when a dense structure retards the dissolving of an implant.

Up to present,  $\alpha$ -TCP are mainly applied in forms of cements and macroporous scaffolds [19], while  $\alpha$ -TCP-based sintered dense blocks have rarely been investigated or reported. Additionally, the stable temperature of  $\alpha$ -TCP (1125-1430 °C) differs from that of  $\beta$ -TCP (< 1250 °C), and HAp is stable in a wide range of temperature between 800 °C and 1400°C [20] according to various published literature data. Hence, it can be assumed that after sintered into dense forms, the properties of these three materials would also differ with each other. Therefore, the purpose of this research is to fabricate dense  $\alpha$ -TCP sintered blocks and investigate the variation of properties under different sintering temperatures, followed by the fabrication of less denser blocks, which have biggest possible porosity meanwhile match with that of cortical bone. Then the samples with both moderate porosity and high compressive strength (> 100 MPa) will be chosen for *in vitro* test in simulated body fluid (SBF), which has ion concentrations nearly equal to those of human blood plasma and was proposed by Kokubo et al. [21], to predict the osteoconduction of the sintered blocks.

## Materials and Methods

### Fabrication of sintered blocks from $\alpha$ -TCP powder by cold isostatic pressing (CIP) method

A cold isostatic pressing (CIP) method [22], was utilized to fabricate the dense sintered blocks.  $\alpha$ -TCP powder ( $\alpha$ -TCP-B, Taihei Chem. Inc., Osaka, Japan) was poured into stainless mold ( $\varnothing=7.00$  mm) and hand-pressed for preforming. The preformed green cylinders were subsequently encapsulated into rubber finger-cot, evacuated and sealed. The sealed green cylinders were then pressed under pressure of 20 MPa in a pre-established water condition for about 30 s. Afterwards, both ends of the green  $\alpha$ -TCP cylinders were rubbed to parallel with the horizontal axis. The received green

cylinders ( $\varnothing=6.15$  mm; length 13.50 mm) were sintered under different conditions and the details are given in Table 1.

### Fabrication of sintered blocks from $\alpha$ -TCP/starch powder by CIP method

Mixture powder containing  $\alpha$ -TCP and potato starch (Potato starch, Nacalai tesque, Inc., Kyoto, Japan) was prepared by milling  $\alpha$ -TCP and starch powders for 20 minutes. Then green cylinders were received by the above-mentioned CIP method and sintered at different temperatures. Detailed information of the fabrication process are listed in Table 2.

### Porosity and shrinkage of the sintered blocks

The shrinkage of the dense sintered blocks were calculated as:

$$\text{Shrinkage (\%)}: s = \left( \frac{V_g - V_s}{V_g} \right) \times 100,$$

Where  $V_g$  is the volume of the green cylinder and  $V_s$  refers to that of sintered block. The volumes of samples were calculated from diameter and length after sintering. Shrinkages of samples are averaged from the values of 6 samples.

**Table 1:** Ion concentrations of simulated body fluid and human blood plasma.

	Ion concentration (mM)	
	Simulated fluid	Blood plasma
Na <sup>+</sup>	142.0	142.0
K <sup>+</sup>	5.0	5.0
Mg <sup>2+</sup>	1.5	1.5
Ca <sup>2+</sup>	2.5	2.5
Cl <sup>-</sup>	147.8	103.0
HCO <sub>3</sub> <sup>-</sup>	4.2	27.0
HPO <sub>4</sub> <sup>2-</sup>	1.0	1.0
SO <sub>4</sub> <sup>2-</sup>	0.5	0.5

**Table 2:** Details of the sintered dense blocks.

Temperature / °C	Heating rate / °C·min <sup>-1</sup>	Time / h	Porosity / %	Shrinkage / %
1150	5	12	19.20±2.3	17.55±1.9
1200	5	12	17.66±1.4	19.90±1.7
1250	5	12	10.85±0.9	24.00±1.6
1300	5	12	9.75±1.1	26.38±1.4
1350	5	12	7.68±0.9	26.28±1.4
1400	5	12	8.30±0.7	28.45±1.5

**Table 3:** Details of the intermediate porous blocks.

Sample No.	Composition of powder / mass %		Heating rate / °C·min <sup>-1</sup>	Keeping time / h	Temperature / °C
	$\alpha$ -TCP	Starch			
91/1300	90	10	5	12	1300
91/1250	90	10	5	12	1250
91/1200	90	10	5	12	1200
91/1150	90	10	5	12	1150
82/1150	80	20	5	12	1150
73/1150	70	30	5	12	1150

The total porosity of all sintered blocks were calculated by Equation:

$$\text{Total porosity (\%)} : \varepsilon = (1 - \rho_b / \rho_s) \times 100,$$

Where  $\rho_b$  is the density of the porous scaffold calculated by weight and volume,  $\rho_s$  is the theoretical density of  $\alpha$ -TCP (2.83g/cm<sup>3</sup>). While apparent porosity of some samples (n=6) were measured through an Archimedes method (according to ASTM C373-88) with kerosene used as a liquid medium and the porosity was expressed as:

$$\text{Apparent porosity (\%)} : \varepsilon = (W_3 - W_1) / (W_3 - W_2) \times 100,$$

Where  $W_1$  is the weight of sample in air,  $W_2$  is the weight of sample suspended in kerosene and  $W_3$  is the saturated weight, and calculated from the density and sample size.

### Immersion test

The 91/1150 samples were soaked in 40 cm<sup>3</sup> SBF solutions for 0, 1, 3, 7 and 14 d, respectively. Composition of SBF with ion concentrations nearly equal to those of human blood plasma as shown in Table 1 [21]. The SBF was prepared by dissolving reagent grade chemicals of NaCl, NaHCO<sub>3</sub>, KCl, K<sub>2</sub>HPO<sub>4</sub>·3H<sub>2</sub>O, MgCl<sub>2</sub>·6H<sub>2</sub>O, CaCl<sub>2</sub> and Na<sub>2</sub>SO<sub>4</sub> into distilled water. It was buffered at pH 7.4 with 50 mmol tris(hydroxymethyl)aminomethane ((CH<sub>2</sub>OH)<sub>3</sub>CNH<sub>2</sub>) and 45 mM hydrochloric acid (HCl) and was kept at 36.5°C. After removal of samples from SBF, the variation of Ca, P and Mg concentrations in the solutions (n=5) were measured by induced coupled plasma atomic emission spectroscopy (ICP-AES; Optima 2000DV, PerkinElmer Japan Co. Ltd., Japan). After soaking, the blocks were dried at 100 °C for 12 h, followed by a vacuum-drying process at 100 C for 24 h to calculate the weight changes before and after soaking.

### Mechanical strength measurement

Compressive strength and Young's modulus of all samples were measured by an Instron Model 5566 system with a crosshead speed of 1 mm/min and the load cell was 10 kN. Each value used in this study was the average data of 6 specimens.

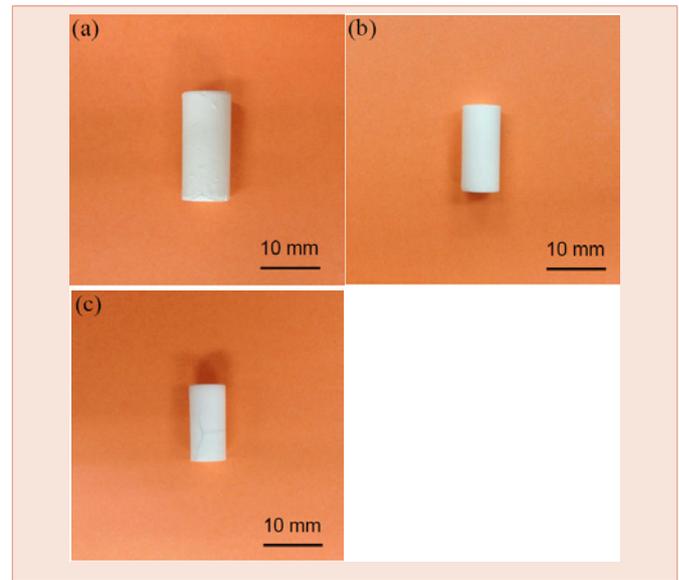
### Crystal phases and morphologies

After sintering process and immersing process, crystal phases of the sintered blocks were tested by powder X-ray diffraction (XRD; RINT2100, Rigaku Co., Japan) using CuK $\alpha$  radiation with a scanning speed of 2°/min at 40 kV and 20 mA after ground into fine powder using a mortar. Morphologies of all the specimens were observed by scanning electron microscopy (SEM; JSM-5600, JEOL Ltd, Tokyo, Japan) at a 15 kV operating voltage after gold sputter-coating.

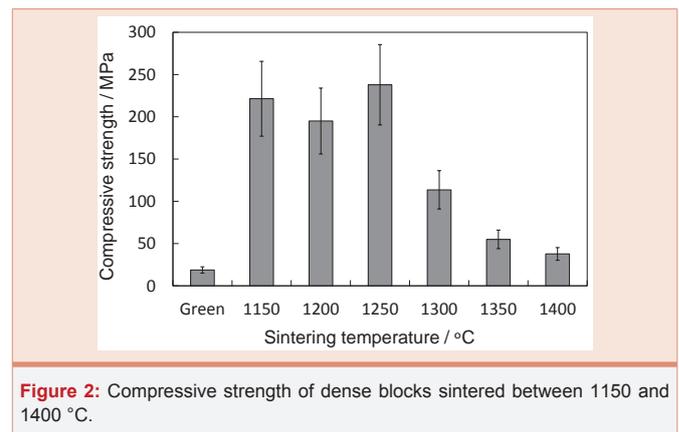
## Results

### Dense sintered blocks prepared from $\alpha$ -TCP

The images of the green cylindrical bodies and sintered blocks are shown in Figure 1. The porosity and shrinkage of the dense sintered blocks are shown in Table 1. It is plain to see that as the sintering temperature increased, the blocks shrank and the porosity dropped from 19.2% to 8.3%, implying that sintered blocks became denser. The compressive strength are given in Figure 2. A turning point of the compressive strength can be found around 1300 °C. The average compressive strength of the blocks sintered between 1150 and 1300



**Figure 1:** Images of green blocks (a) and sintered dense blocks. (b) 1150 °C, (c) 1400 °C.



**Figure 2:** Compressive strength of dense blocks sintered between 1150 and 1400 °C.

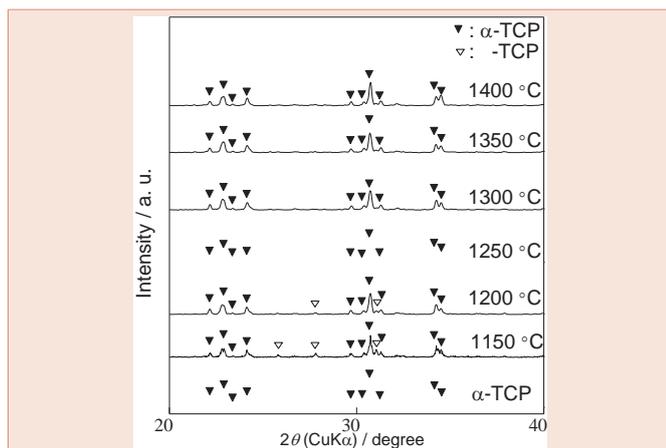
°C exceeded 100 MPa, however, decreased precipitously at 1300 °C—from about 230 MPa to 110 MPa, and fell further down below 100 MPa when the temperature came to 1350 °C.

After sintered at different temperature varying from 1150 °C to 1400 °C, the crystal phase of the samples were analyzed by XRD (Figure 3). In samples sintered between 1250 °C and 1400 °C, no other than peaks corresponding to  $\alpha$ -TCP were detected. Although  $\alpha$ -TCP phase was the main phase, there were still small amount of peaks corresponding to  $\beta$ -TCP presented in the samples sintered at 1150 °C and 1200 °C. Through the surface SEM images (Figure 4 b,c,d) of the sintered blocks, it can be observed that the grains grew as the temperature went higher, from about 3  $\mu$ m up to around 15  $\mu$ m. The cross-section SEM images (Figure 4e,f,g) showed a proceeding densification in the sintering process. With the sintering temperature increasing, the quantity of the pores declined and the interconnected pores gradually became isolated, which matched the tendency of the porosity.

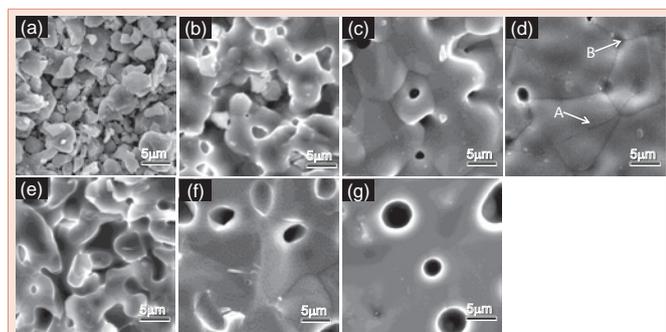
## Dense sintered blocks prepared from $\alpha$ -TCP and starch

Green cylinders containing 90 mass%  $\alpha$ -TCP and 10 mass% were sintered between 1150 °C and 1300 °C. The total porosity and apparent porosity were measured and shown in Figure 5. The porosity ranges between 22% and 35%, in which apparent porosity was slightly, about 3% lower than total porosity, indicating that most pores are interconnected. The highest compressive strength was obtained in the 91/1300 samples yet those of other samples were also successfully maintained above 100 MPa. As the content of starch rose up to 20 mass%, the compressive strength dropped quickly below 100 MPa. In the cross-section SEM images of the sintered blocks (Figure 6a,b,c), pores (20  $\mu$ m ~ 80  $\mu$ m) caused by starch could be observed and were interconnected through the naturally generated pores (around 5  $\mu$ m).

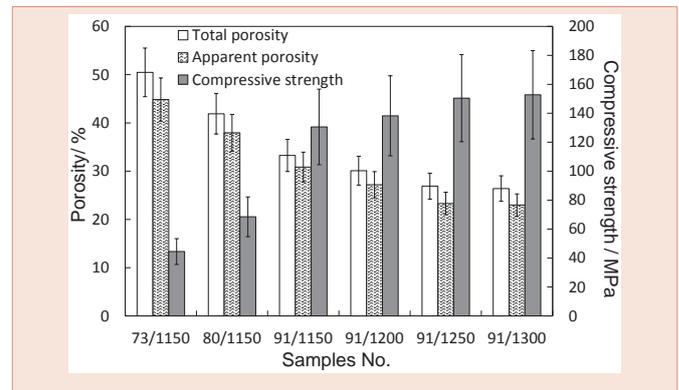
The 91/1150 group samples were soaked in SBF solutions for 1, 3, 7 and 14 d for *in vitro* evaluation. After immersing and drying process, each sample gained weight, as shown in Figure 7, and the extra weight increased along with immersion period. As for the compressive strength, it continued dropping downward in the initial 7 days and did not recover until 14 d, yet still lower than the initial strength. The changes in crystalline phases after immersion are



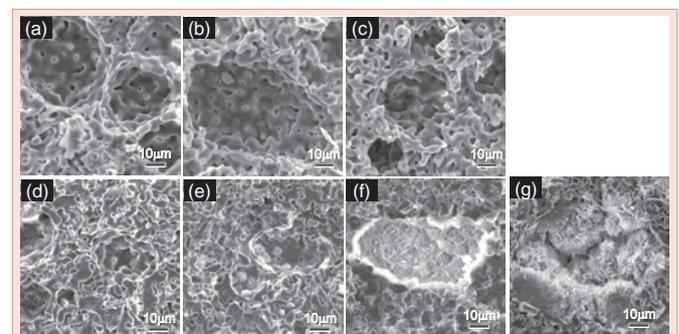
**Figure 3:** XRD patterns of starting powder  $\alpha$ -TCP and samples after sintering between 1150 and 1400 °C.



**Figure 4:** SEM images of green blocks (a) and dense sintered blocks. Surface: (b) 1150 °C, (c) 1300 °C, (d) 1400 °C. Cross-section: (e) 1150 °C, (f) 1300 °C, (g) 1400 °C.



**Figure 5:** Total porosity, apparent porosity and compressive strength of the intermediate porous sintered blocks.



**Figure 6:** Cross-section SEM images of the sintered intermediate porous blocks. (a) 73/1150, (b) 82/1150, (c) 91/1150. (d), (e), (f) and (g) are cross-section SEM images of 91/1150 samples after soaking for 1, 3, 7 and 14 d, respectively.

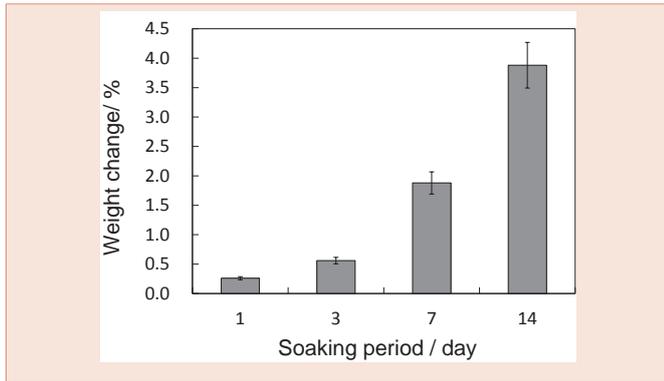
given in Figure 9. With soaking periods extending, the pre-existing peaks corresponding to  $\beta$ -TCP gradually strengthened. Conversely, those of  $\alpha$ -TCP weakened and some of the peaks even disappeared. In the 14 d samples, peaks corresponding to HAp were detected. The morphology of the *in vitro* tested samples are shown in Figure 8. Except for the 1 d soaked samples, newly formed crystalline were observed in all samples, mainly in and around the pores. Relatively more new crystalline were observed in 7 d and 14 d samples.

Ca, P and Mg concentrations in SBF solutions before and after the immersion test were measured by ICP-AES and shown in Figure 10. Ca concentration in SBF slightly and gradually increased up to 14 d. On the contrary, the P and Mg concentrations showed analogous downward tendency but the declining rates were faster than the increase rate of Ca concentration.

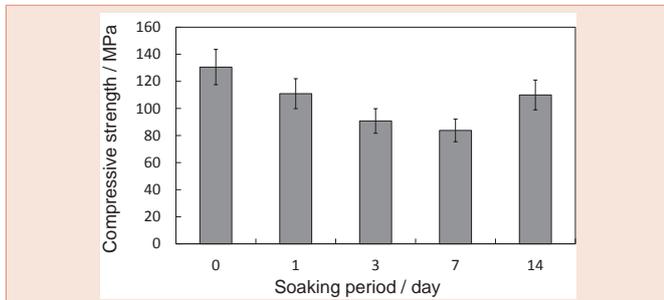
## Discussion

### Dense sintered blocks prepared from $\alpha$ -TCP

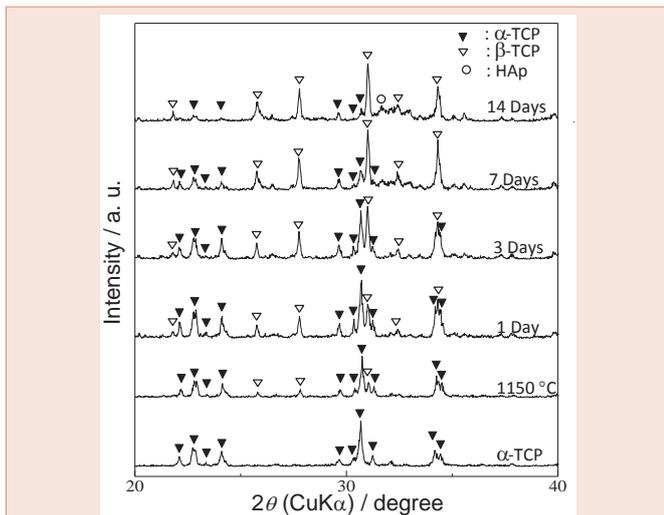
The porosity of the dense sintered blocks falls as the sintering temperature increases, implying a proceeded sintering process. From the surface SEM images (Figure 4b,c,d) it can be seen that the isolated grains in the green cylinders gradually merge with adjacent grains and grain sizes also grow from about 3.00  $\mu$ m to around 15.00  $\mu$ m.



**Figure 7:** Weight changes of 91/1150 samples after soaking for 1, 3, 7 and 14 d, respectively.



**Figure 8:** Compressive strength variation of 91/1150 samples before and after soaking for 1, 3, 7 and 14 d, respectively.



**Figure 9:** Crystalline phases changes of 91/1150 samples before and after soaking for 1, 3, 7 and 14 d, respectively.

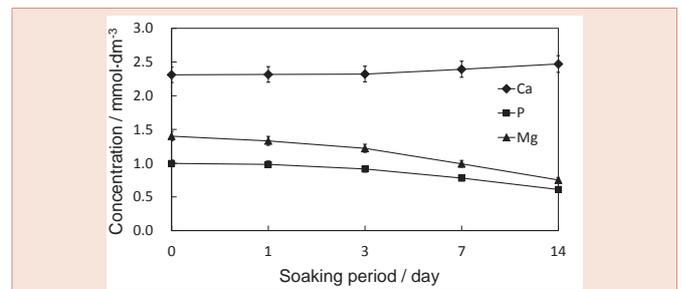
Together with the cross-section SEM images (Figure 4b,c,d) ,it can be observed that sintering process is accompanied by densification of the blocks, during which the amount of pores goes down and the channels of the interconnected pores slowly collapsed to become

isolated. However, the compressive strength, as shown in Figure 2, does not increase with the proceeding sintering process. Although the compressive strength of four groups (1150, 1200, 1250 and 1300 °C) exceeds the minimum value (100 MPa) of human cortical bone, the compressive strength drops dramatically at 1300 °C, from 230 MPa to 110 MPa, and further decrease below 100 MPa when temperature become higher than 1350 °C. A. Tricoteaux et al. [23], reported a  $\beta$ -TCP slip (porosity: 0 ~ 14%), and the mechanical strength increased as the porosity decreased. The results of this research, in contrary, show a opposite tendency. This unexpected phenomenon could be attributed to internal stresses. A common source of internal stresses in ceramic materials is the thermal expansion anisotropy. In some cases residual stresses are large enough to produce microcracking along grain boundaries or macrocrack propagation [24].

In this research, macrocracks are frequently observed on the blocks sintered at 1350 °C and 1400 °C, as shown in Figure 1c). It can be deduced that in the preservation stage of sintering process, the temperature of the cylinder surface is same as that of the center and thermal stress is not generated. During the cooling stage, the surface cools faster and contracts more than the center, bringing about tensile stress on the surface and produce compressive stress in the center. When the tensile stress is large enough and exceeds the tensile strength, macrocracks are usually produced. As can be seen in Figure 4, the sintering process does not proceed optimally, during which the grains grow inhomogeneously and microscopic stress may occur between and in the adjacent grains, leading to the microcracks along the grain boundary (boundary crack, arrow A), and some grains are directly broken by this microscopic stress (cleavage crack, arrow B). In addition, the size of these two types of microcracks is positively correlative with the size of the grain, implying that the cracks propagate with the growth of the grain. The relation between the fracture strength and grain size of a ceramic material is researched by S. Mizuta et al. [25], and expressed as an empirical equation:

$$\sigma_f = \sigma_0 + kd^{-1/2}$$

Where  $\sigma_f$  is the fracture strength, d is the grain size,  $\sigma_0$  and k are constants. It can be deduced from this equation that a larger sized grain can engender negative effects on the mechanical strength of a dense sintered cylinder. When the cylinder is compressively pressed, the above-mentioned macro and micro cracks both propagate, thus rendering the entire failure of the blocks. Meanwhile, relatively higher compressive strength exhibited in the cylinders sintered



**Figure 10:** Concentrations of Ca, P and Mg in SBF solution after soaking 91/1150 samples for 1, 3, 7 and 14 d, respectively.

between 1150 °C and 1250 °C may be ascribed to the higher porosity. The interconnected pores provide spaces for the grains to deform and migrate towards pore spaces instead of being enclosed and extruded by adjacent grains, conducing to release the thermal and microscopic residual stresses.

### Dense sintered blocks prepared from $\alpha$ -TCP and starch

In order to increase the porosity of the sintered blocks to match with that of human cortical bone,  $\alpha$ -TCP and starch are mixed for sintering. Sintering temperature are controlled between 1150 °C and 1300 °C, through which pores are interconnected and compressive strength is above 100 MPa. Both total porosity and apparent porosity are measured and shown in Figure 5. The porosity of the 91/1150 samples is exactly a little bit higher than that of human cortical bone. Apparent porosity is slightly lower than total porosity, indicating that most pores are interconnected and may provide spaces for circulation of nutrients and exhaust of metabolic waste. The increase of the porosity is accompanied with different degrees of strength reduction, yet except the 73/1150 and 82/1150 samples, the compressive strength of other samples are maintained above 100 MPa. The naturally generated pores ( $< 5 \mu\text{m}$ ) in the sintering process are generally deemed to be difficult for cells to attach and migrate, but useful for the protein attachment and cell adhesive [26]. In this research, the starch mixed with  $\alpha$ -TCP not only increase the porosity, but also produce pores with sizes ranging from 30 to 90  $\mu\text{m}$  (Figure 6a,b,c). The size of these pores are comparable to that of human average cell size (74.66 ~ 82.45  $\mu\text{m}$ ) [27], especially for that of osteoclast (20 ~ 150  $\mu\text{m}$ ) [28], which will be suitable for osteoclast to migrate in the pores and resorb the bone-like materials, thus leaving space for osteoblast (10 ~ 50  $\mu\text{m}$ ) [29], to make new bone tissues.

Taking both compressive strength and porosity into account, the 91/1150 samples, which have the most porous structure in the above-100 MPa samples, are chosen for the *in vitro* immersion test in SBF solutions up to 14 d. A possible reason that each sample gains weight after soaking is the newly formed crystalline, as shown in (Figure 6d,e,f,g), and the amount of observed crystalline is proportional to the extra weight. The compressive strength of samples keep declining up to 7 d and do not rebound above 100 MPa until 14d, yet lower than the initial compressive strength. As immersion proceeds, crystalline phases of the samples change, but not as expected. There are no peaks corresponding to HAP being detected until the 14 d meanwhile  $\alpha$ -TCP peaks gradually weaken and by 14 d the  $\beta$ -TCP become the main phase. Putting this phase change and the weight change together into consideration it can be presumed that  $\alpha$ -TCP dissolve in SBF solution and the released  $\text{Ca}^{2+}$  and  $\text{PO}_4^{3-}$  along with the pre-existing  $\text{Ca}^{2+}$  and  $\text{HPO}_4^{2-}$  ( $\text{HPO}_4^{2-} \rightarrow \text{H}^+ + \text{PO}_4^{3-}$ ) in SBF recrystallize and form  $\beta$ -TCP. To confirm this assumption, Ca and P concentrations in SBF are measured and should simultaneously decline as expected. However, as can be seen in Figure 10, only concentration of P decreases and contrarily concentration of Ca slightly increases. In regard to this phenomenon, we paid attention to a  $\beta$ -TCP analogue, referred as Mg-whitlockite ( $\text{Mg-}\beta$ -TCP,  $(\text{Ca}, \text{Mg})_3(\text{PO}_4)_2$ ), in which calcium is partly substituted by magnesium. Mg-whitlockite has been commonly found in biological

conditions as well as fabricated for research [30,31]. For this reason the concentration of Mg is subsequently measured and the result shows that concentration of Mg in SBF presents a similar downward tendency with P, indicating that the substitution of Ca by Mg actually occurs in the immersion process and contributes to the increase of Ca concentration.

### Conclusion

Both dense and intermediate porous  $\alpha$ -TCP blocks were fabricated through cold isostatic pressing followed by a sintering process (1150 ~1400 °C). The dense blocks reached a highest compressive strength (230 MPa) after sintering at 1250 °C and it dropped steeply with higher sintering temperature. The decrease in compressive strength might be attributed to the thermal residual stress, the microscopic residual stress between grains and the growth of the grains. Intermediate porous blocks (compressive strength 135 MPa, porosity 33%) were soaked in simulated body fluid for *in vitro* test. The compressive strength kept decreasing at first and by 14 d it rebounded to around 110 MPa.  $\alpha$ -TCP gradually dissolved and transformed into  $\beta$ -TCP, in which Ca was partially substituted by Mg in SBF solution. HAP formation was not observed until the soaking process lasted for 14 d.

### Acknowledgements

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